

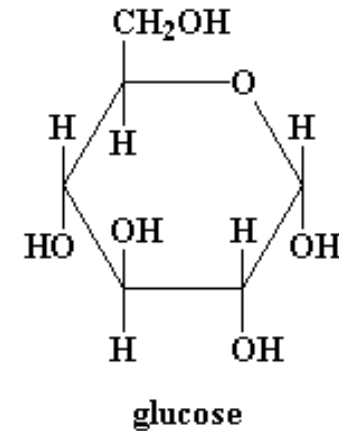
Carbohydrate metabolism

The name carbohydrate arises from the basic molecular formula



$(\text{C} + \text{H}_2\text{O})_n$ hydrates of carbon

$n=3$ or more.



Energy from the sun captured by green plants during photosynthesis is stored in the form of **carbohydrates**.

Carbohydrates are the metabolic precursors of virtually all other biomolecules.

Source of energy

Phototroph: an organism that obtains energy from sunlight for the synthesis of organic compounds (they convert the solar energy to chemical one)

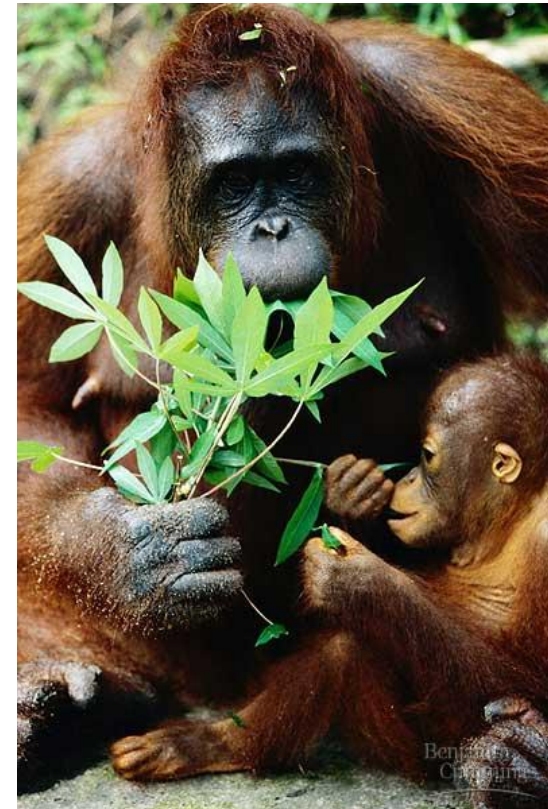
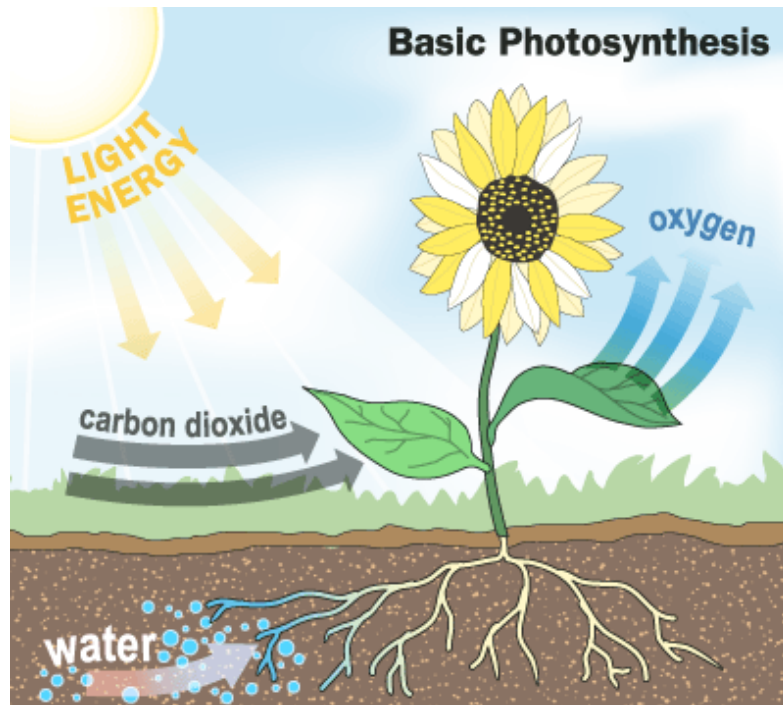
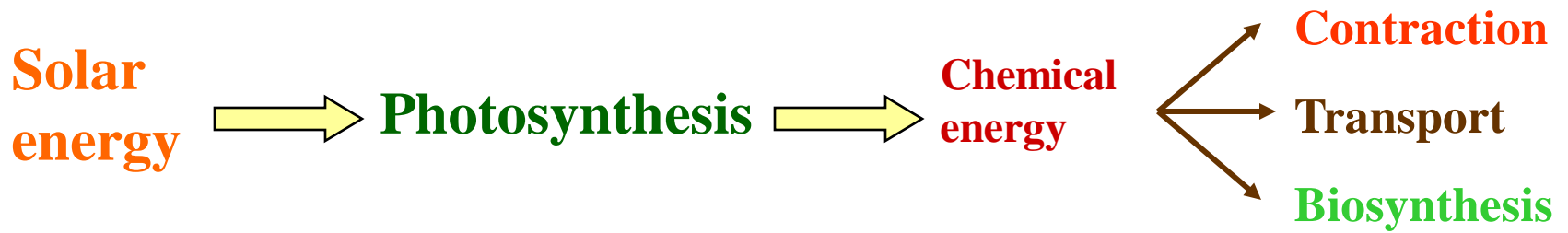
Chemotroph: an organism that cannot harvest and convert the solar energy, instead of it take up organic compounds and oxydize them to gain energy.

Source of carbon

Autotroph: An organism capable of synthesizing its own food from inorganic substances, using light or chemical energy. Green plants, algae, and certain bacteria are autotrophs.

Heterotroph: An organism that cannot synthesize its own food and is dependent on complex organic substances for nutrition.





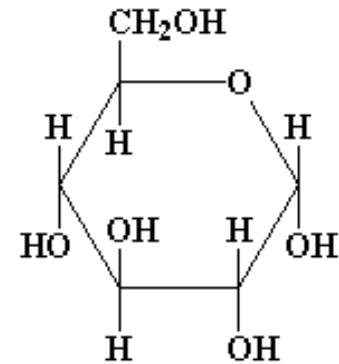
Breakdown of carbohydrates provides the energy that sustains animal life.

Monosaccharides: cannot be broken down into smaller sugars under mild conditions

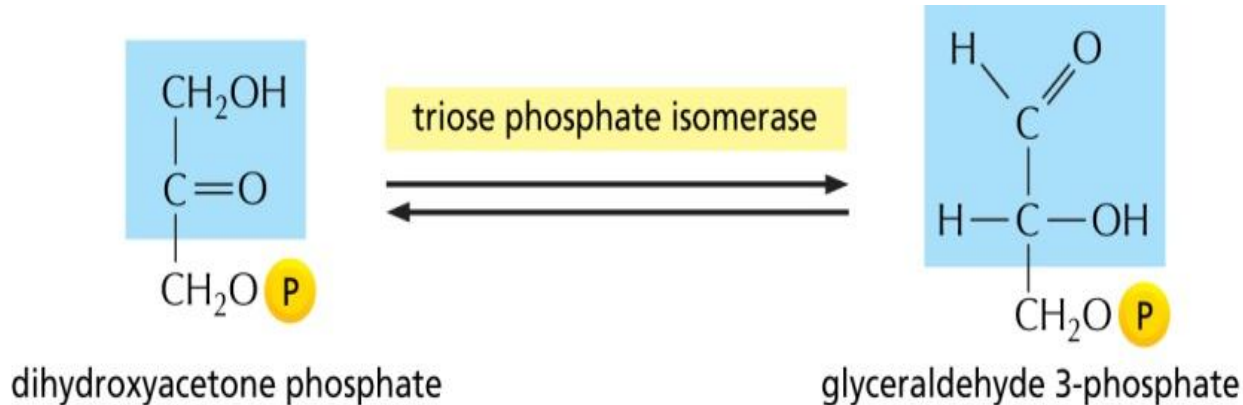
Consist typically of three to seven carbon atoms

Aldoses: aldehyde function group or a

Ketoses: ketone function group



glucose

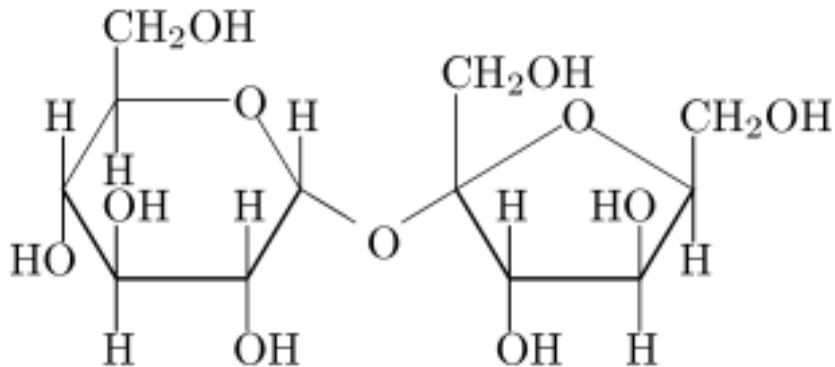
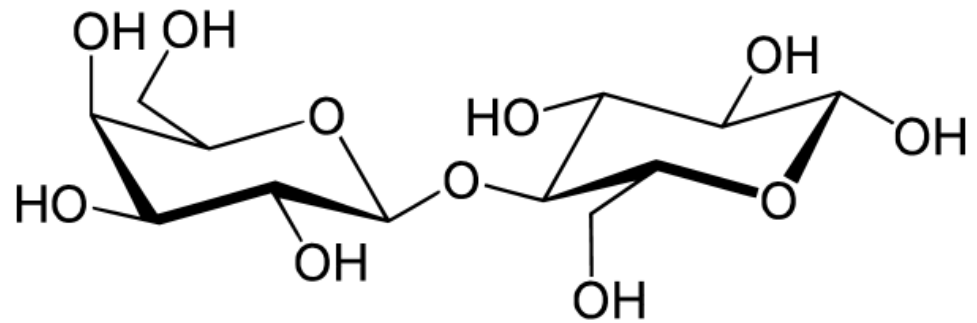


The simplest monosaccharides are water-soluble, and most taste sweet.

Oligosaccharides: consist of from two to ten simple sugar molecules

Oligo: Greek word meaning “few”

Disaccharides are common in nature consist of two monosaccharide units linked by a **glycosidic bond**.



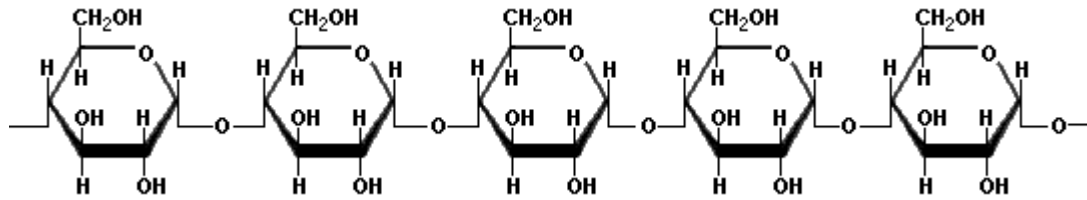
sucrose

lactose

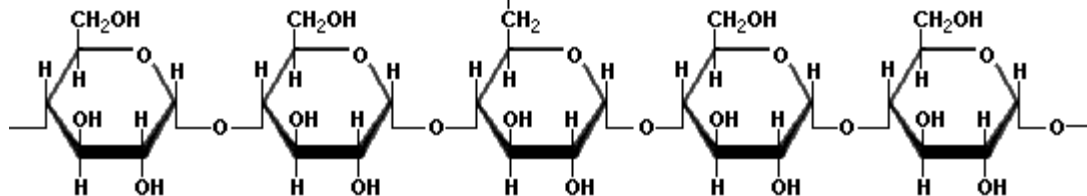
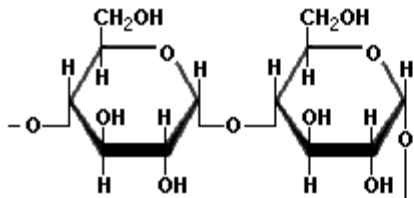
Polysaccharides: are polymers of the simple sugars and their derivatives

They may contain **hundreds or even thousands of monosaccharide units**. Their molecular weights range up to 1 million or more.

linear or branched polymers



amylose



amylopectin



Complex amylopectin structure

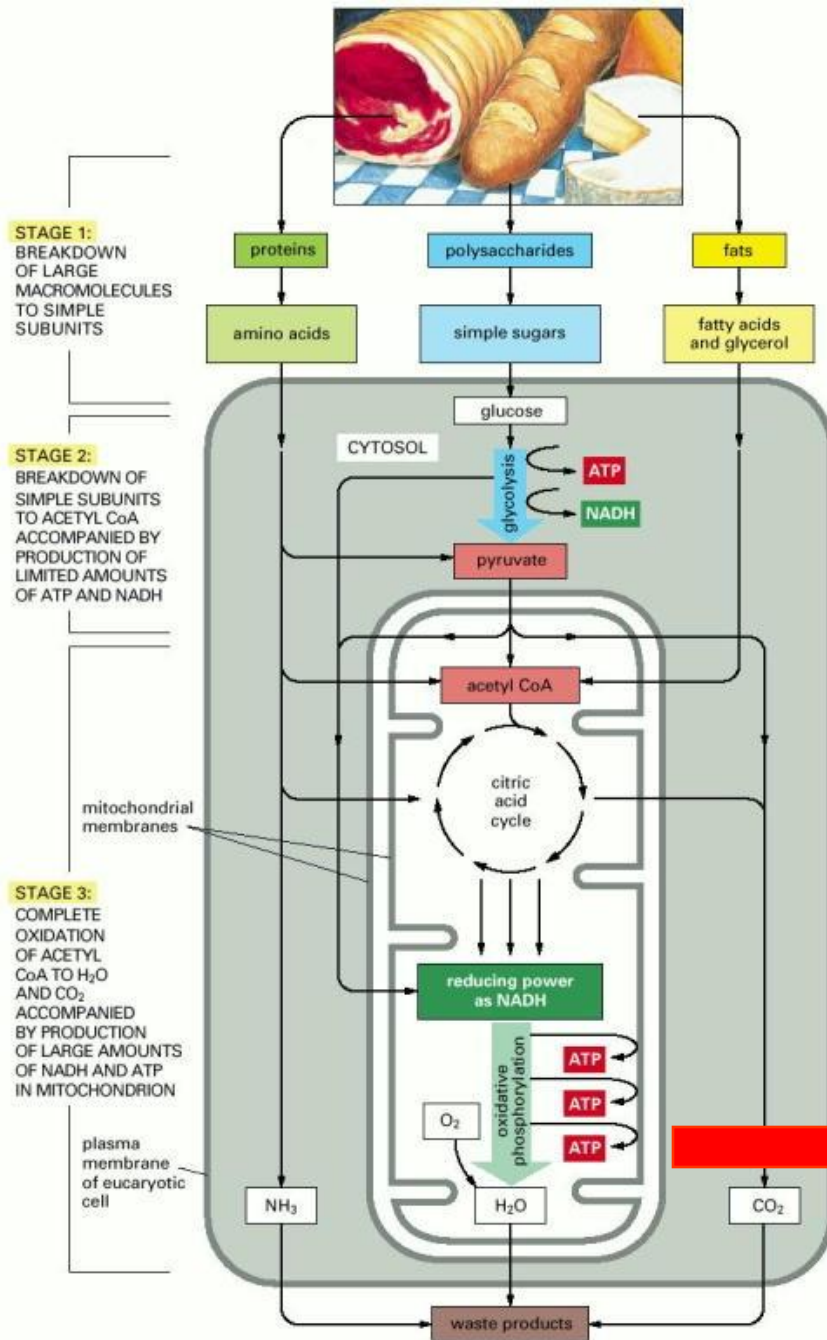
Polysaccharides are storage materials, structural components, or protective substances.

Starch, glycogen: provide energy reserves for cells
organisms store carbohydrates in the form of polysaccharides rather than as monosaccharides to lower the osmotic pressure of the sugar reserves

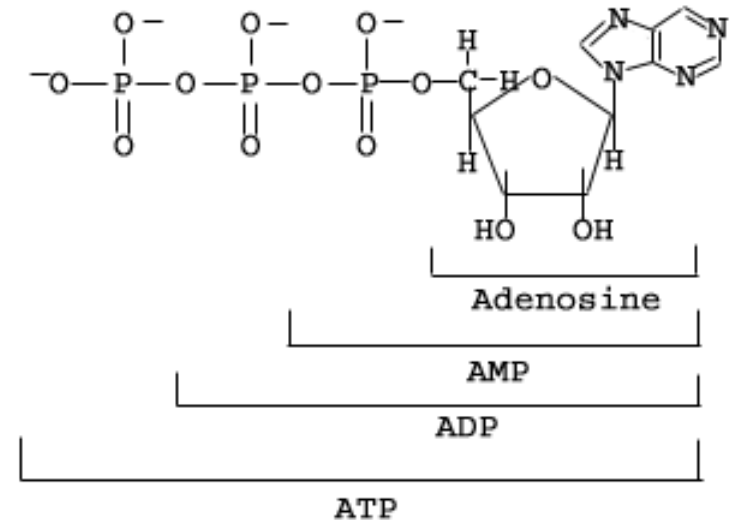
Mucopolysaccharides: eg. the hyaluronic acids, form protective coats on animal cells.

Chitin, cellulose: provide strong support for the skeletons of arthropods and green plants

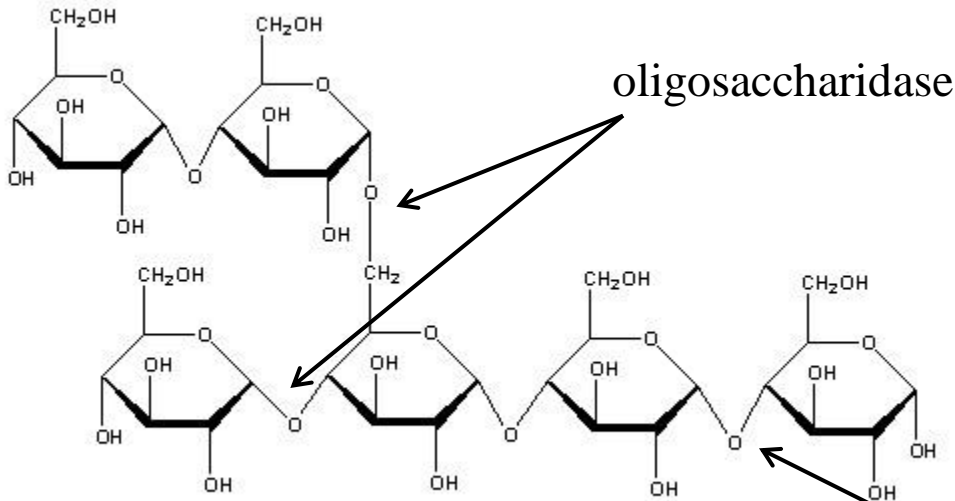
The polysaccharides (together with proteins, lipids) must be **broken down into smaller molecules (monomers)** before our cells can use them either as a source of energy or as building blocks for other molecules.



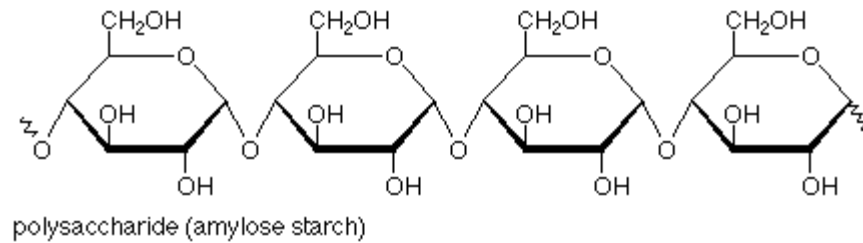
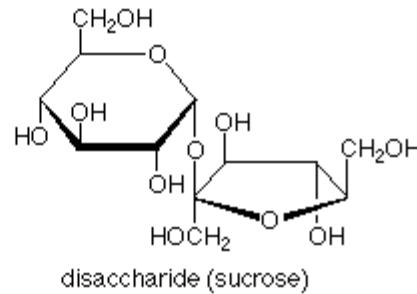
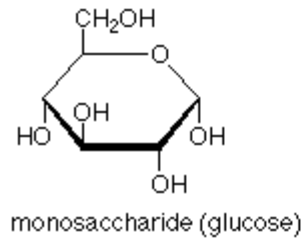
Energy currency: **ATP**



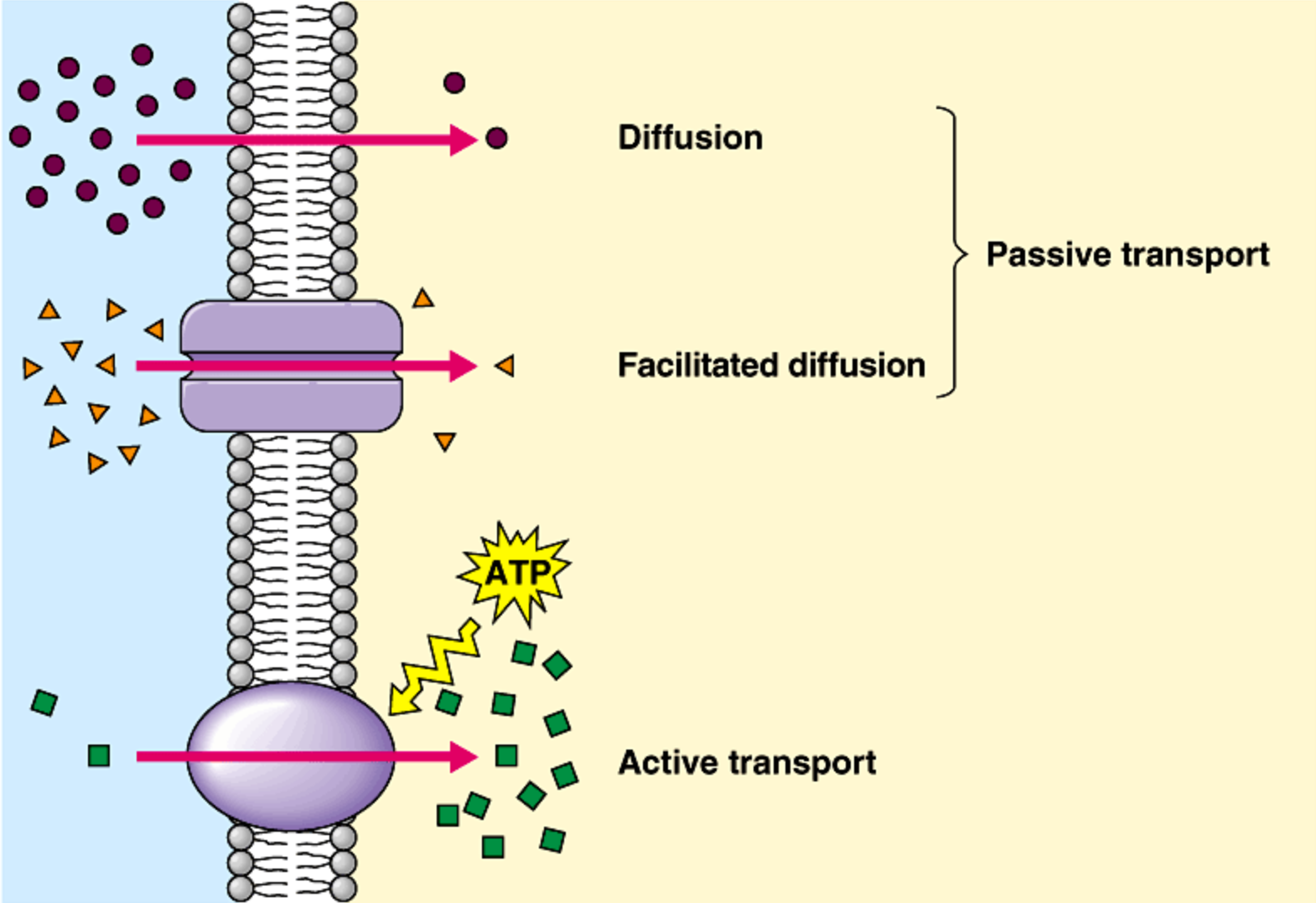
starch



α -amilase



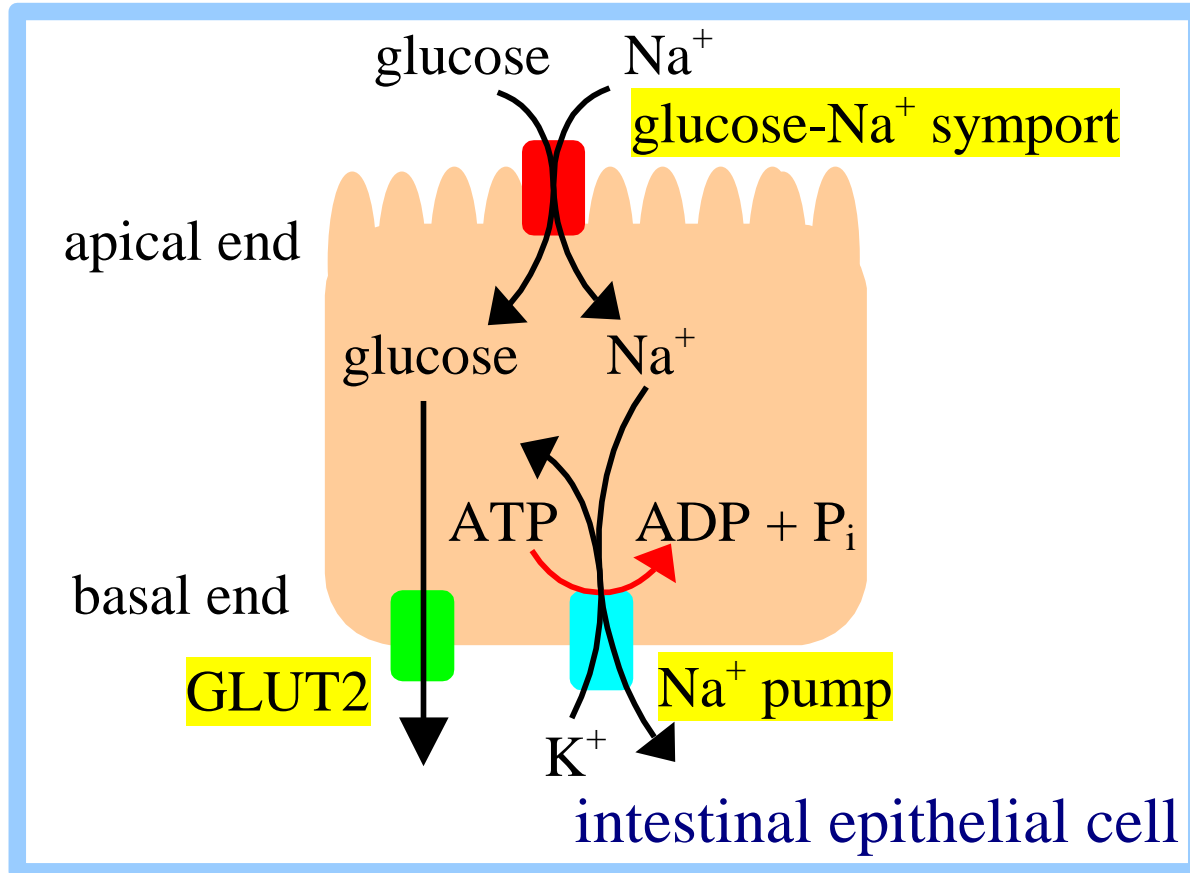
Simple diffusion: the compounds permeate freely through the membrane to the direction of concentration gradient. Quite rare.



Transport of glucose

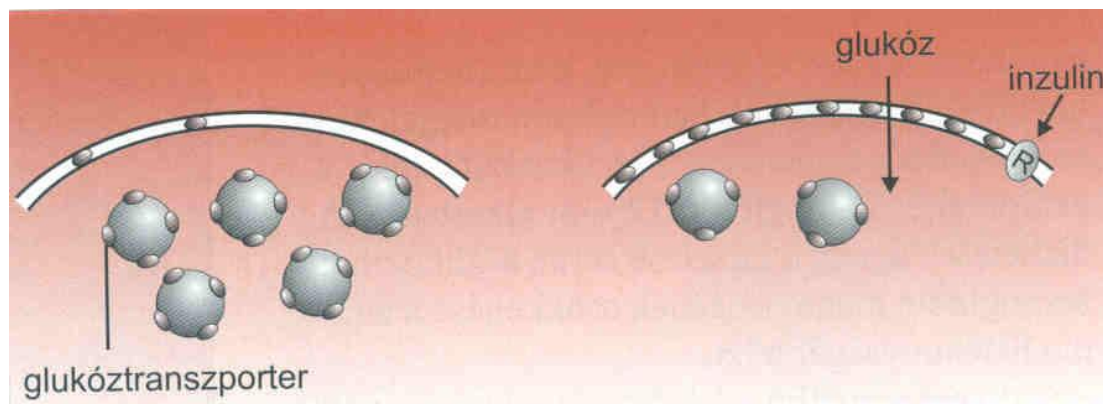
The uptake of glucose from the lumen of intestine is coupled to the uptake of Na^+ . The driving force is the electrochemical gradient of Na^+ .

The electrochemical gradient of Na^+ has been built up by the Na^+/K^+ pump on the expense of ATP hydrolysis.

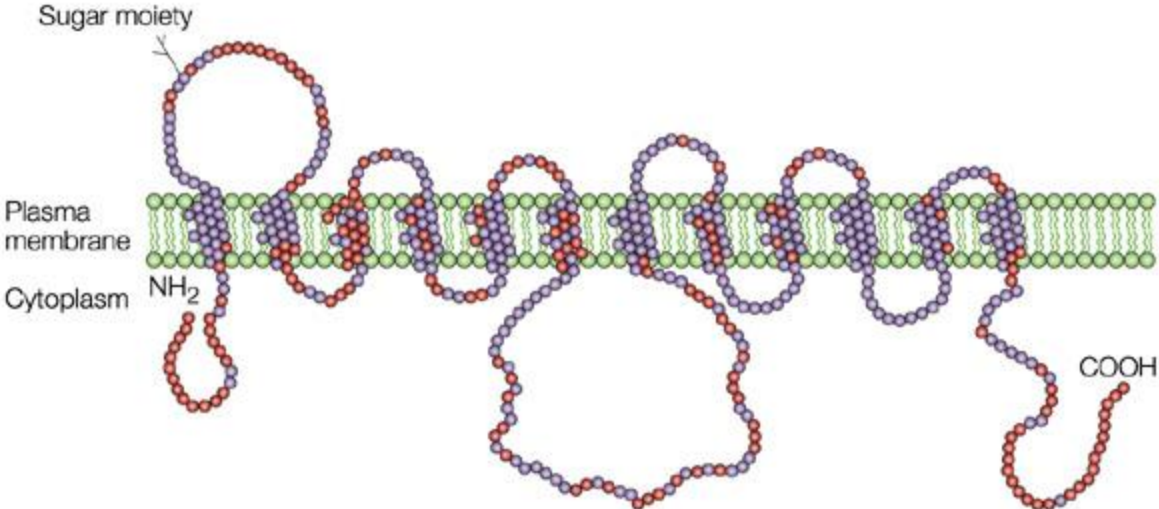


Glucose transporters (GLUT family)

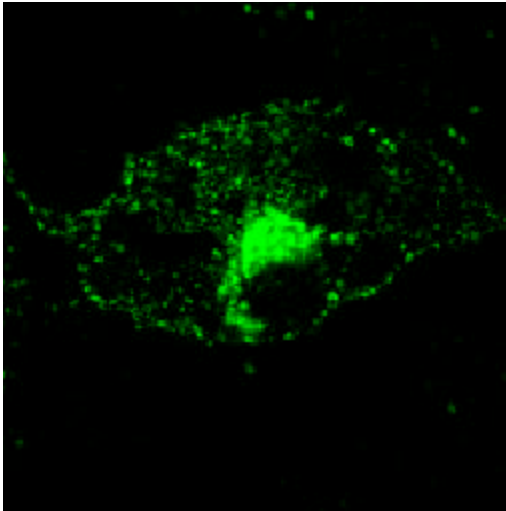
- GLUT 1: red blood cell, brain, muscle adipose tissue, insulin independent
- GLUT 2: liver cells, pancreatic b-cell, kidney cells, intestinal epithelium, high K_m value
- GLUT 3: nerve cells, low K_m value
- GLUT 4: muscle, adipose tissue, insulindependent
- GLUT 5: fructose transporter



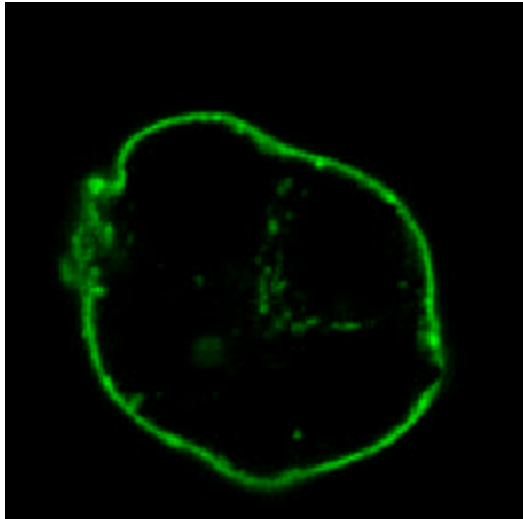
The structure of GLUT4



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insulin



Glycolysis

Glycolysis is occurred in all human cells. Glucose is the central fuel of metabolism. All cells can utilize it.

glykys = sweet, lysis = cleavage

Daily glucose demand of the human body: ca. 160 g

central nerve system, brain: 120 g

ATP- synthesis: 40 g

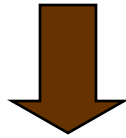
ATP is generated in anaerobic conditions

The reactions of glycolysis

The first discovered metabolic pathway

All reactions are localized to the cytosol

The enzymes are organized into **multienzyme complexes**



The intermediates are channeled from one enzyme to the other

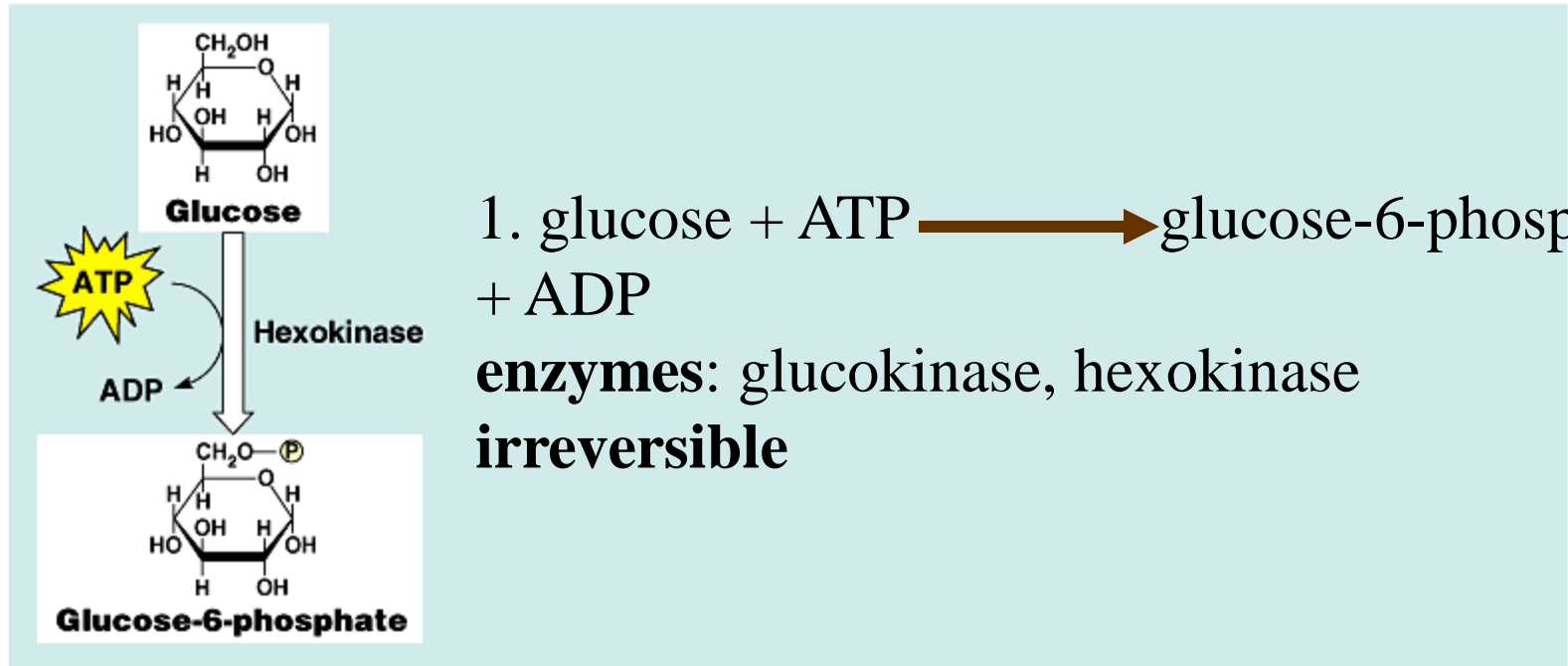
All intermediates are phosphorylated



The cell membrane is **not permeable** for them

Reversible and irreversible reactions

The first six C phase

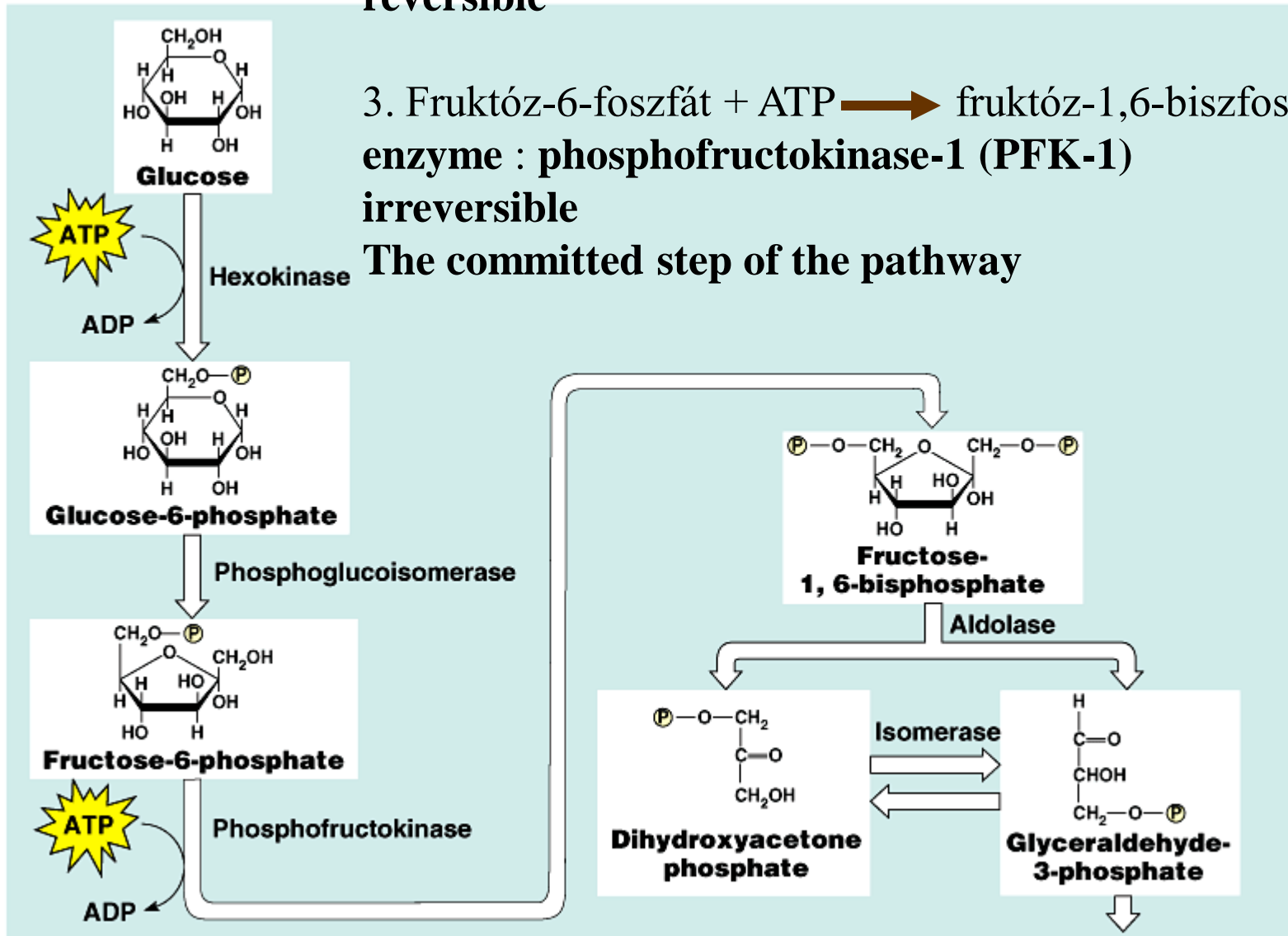


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glucose-6-P cannot be transported back across the plasma membrane

2. glucose-6-phosphate \longrightarrow fructose-6-phosphate
 enzyme: phosphoglucose isomerase
 reversible

3. Fructóz-6-foszfát + ATP \longrightarrow fruktóz-1,6-biszfoszfát + ADP
 enzyme : phosphofruktokinase-1 (PFK-1)
 irreversible
 The committed step of the pathway



4. fructose-1,6-bisphosphate

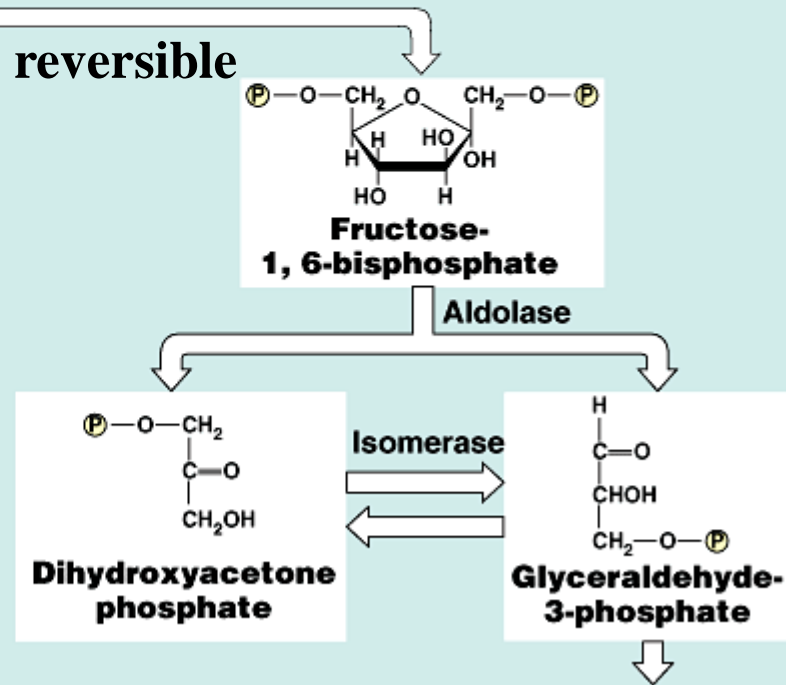
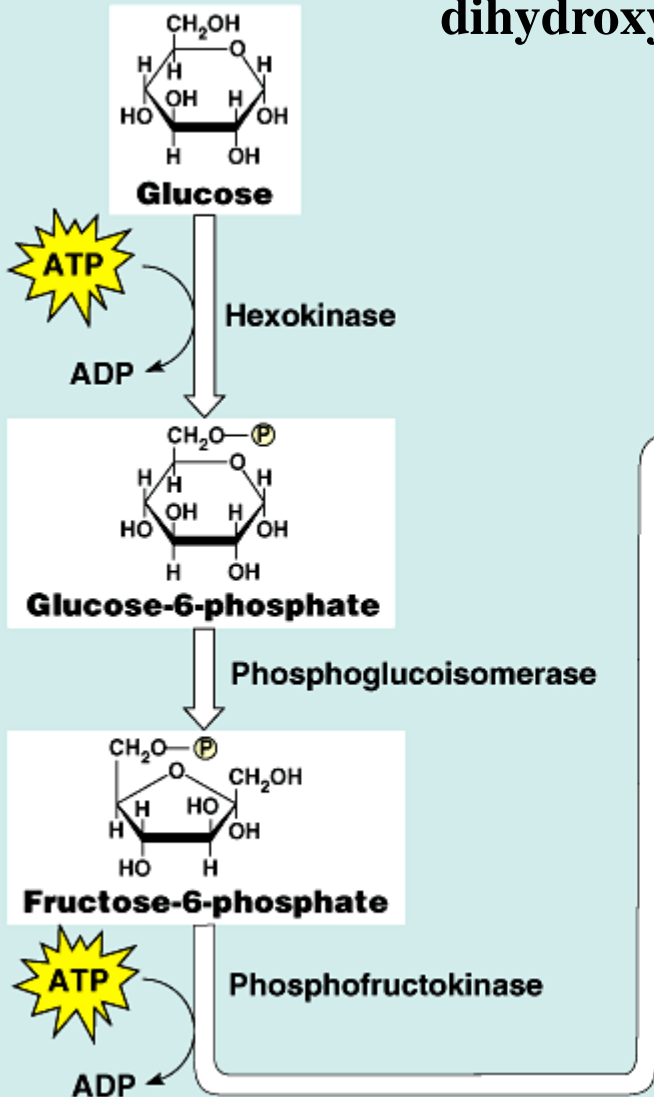
dihydroxyacetone phosphate glyceraldehyde 3-phosphate

enzyme: aldolase reversible

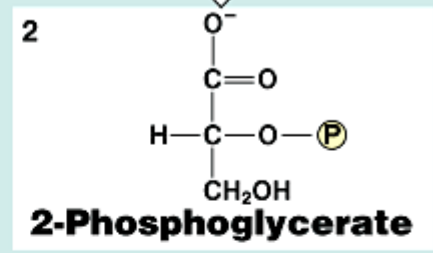
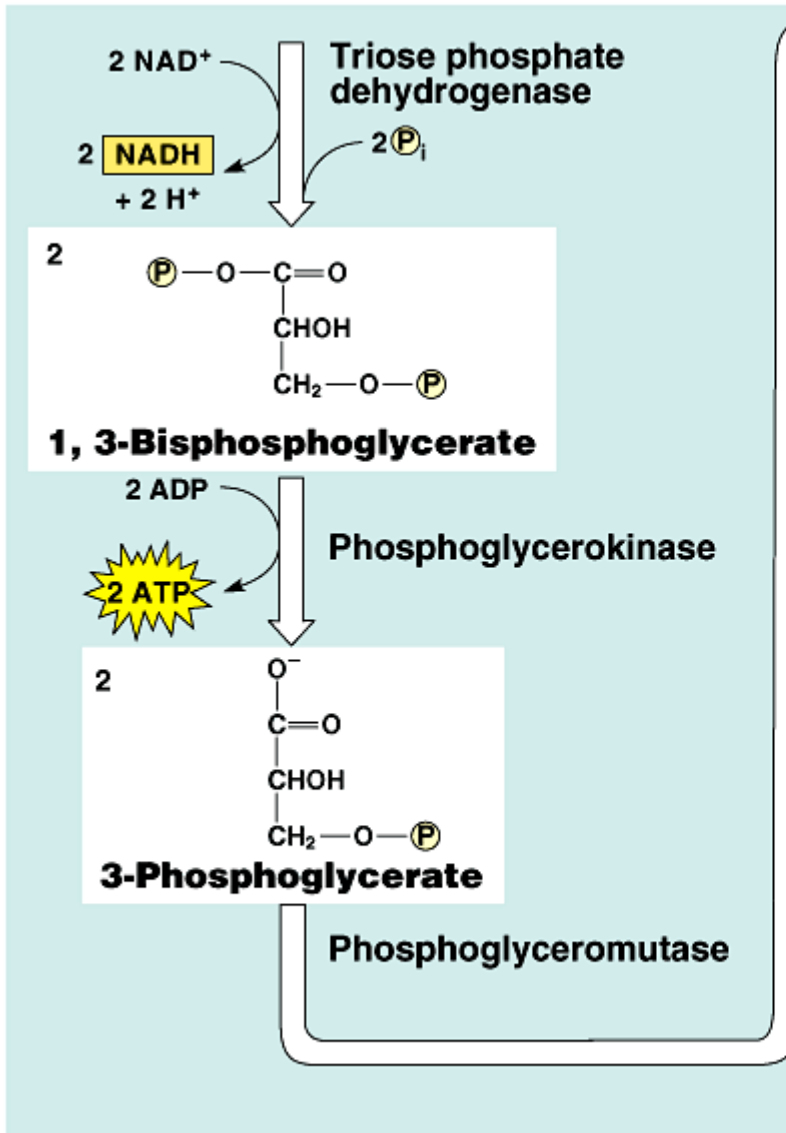
5. Triose-phosphates can convert to each other

Enzyme: triose phosphate isomerase

reversible



The second 3 C phase



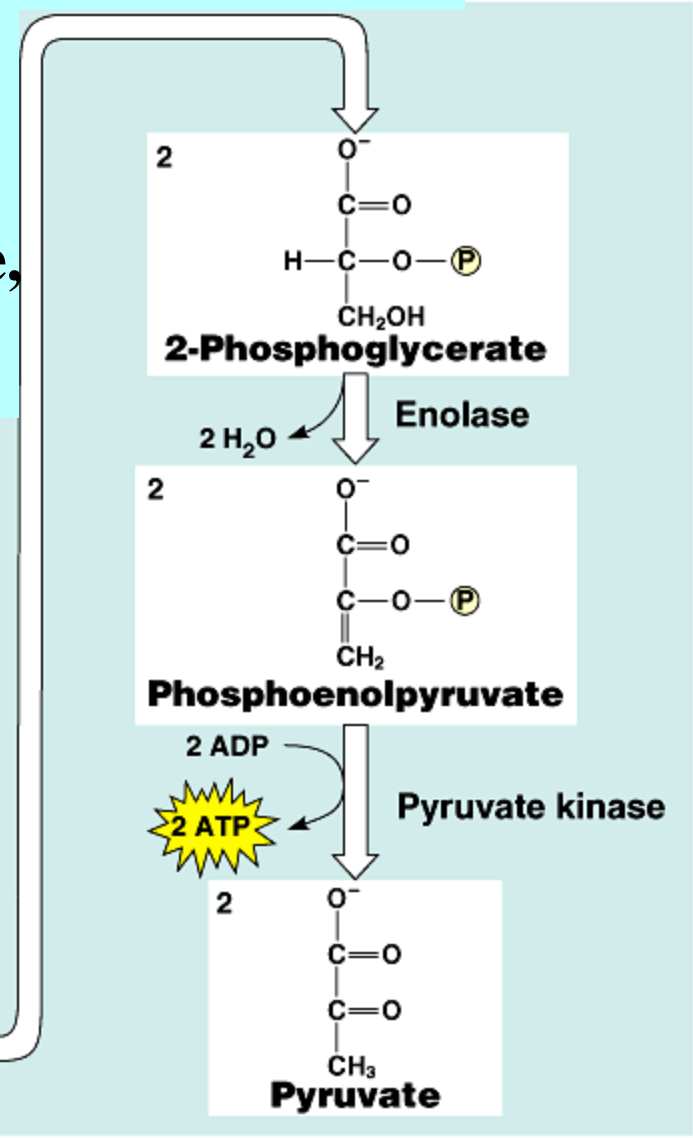
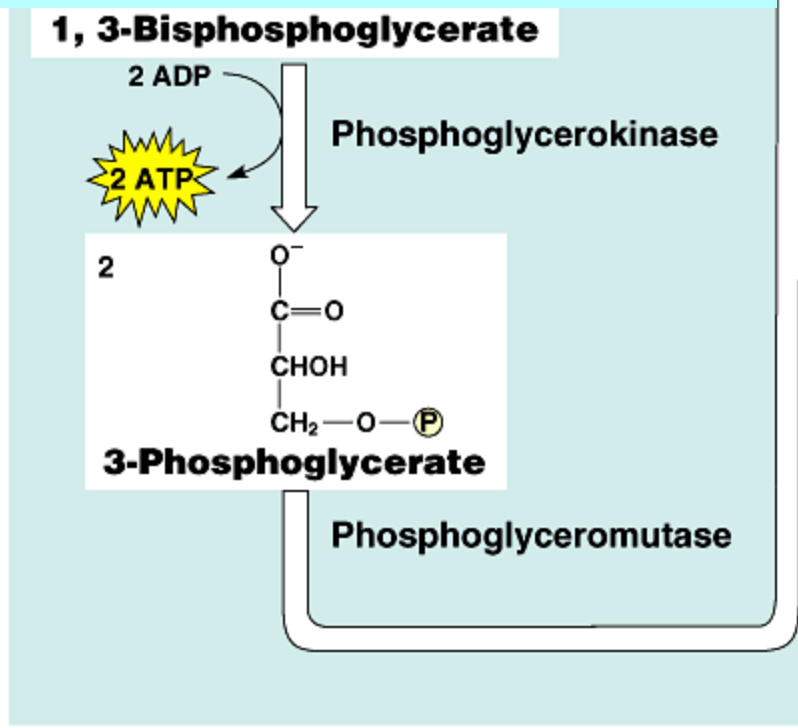
6. glyceraldehyde 3-phosphate → 1,3-bisphosphoglycerate
 enzyme: glyceraldehyde-3-P dehydrogenase
 reversible
7. 1,3-bisphosphoglycerate + ADP → 3-phosphoglycerate + ATP
 enzyme : phosphoglycerate kinase, reversible
 substrate-level phosphorylation
8. 3-phosphoglycerate → 2-phosphoglycerate
 enzyme : phosphoglycerate mutase, reversible

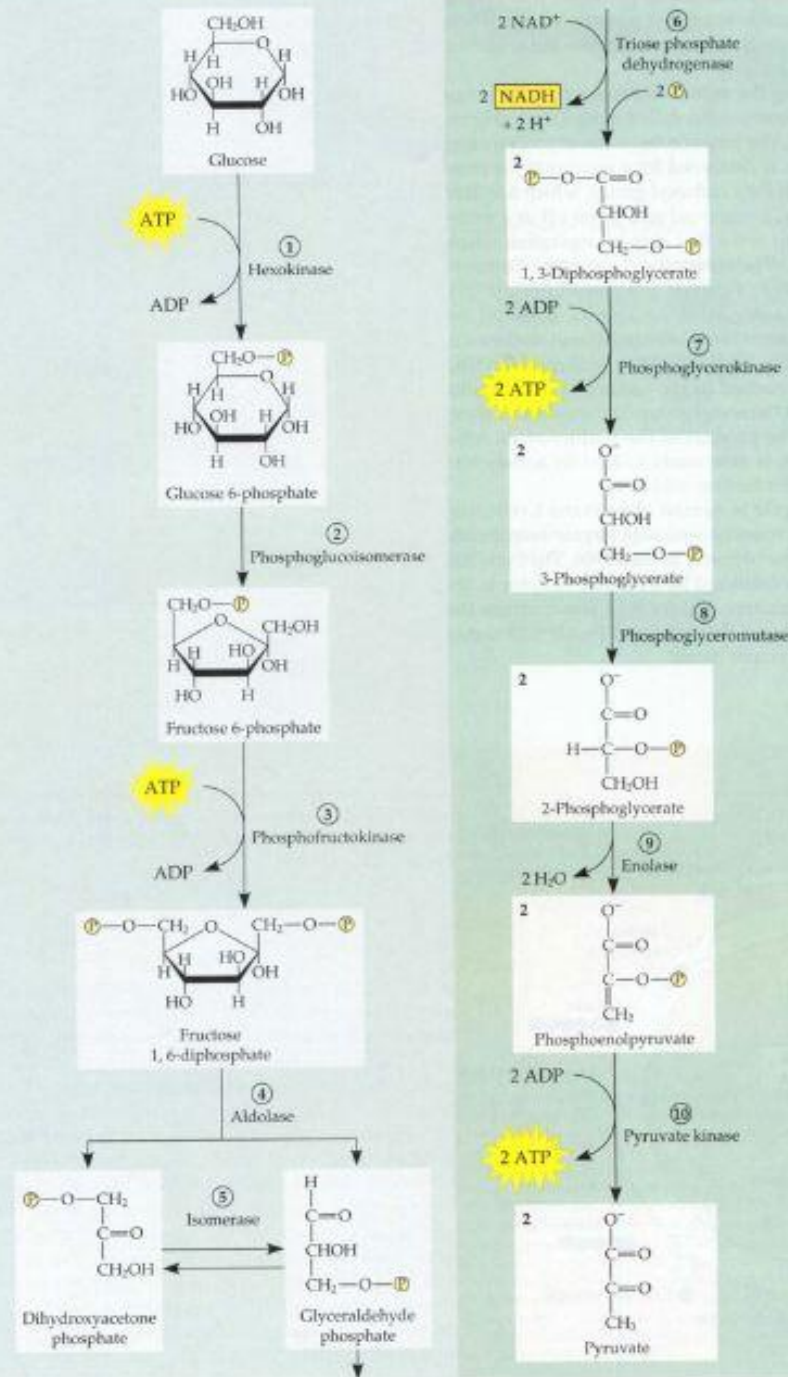
9. 2-phosphoglycerate \longrightarrow phosphoenolpyruvate

enzyme: enolase, reversible

10. phosphoenolpyruvate + ADP \longrightarrow pyruvate + ATP

enzyme: pyruvate kinase, irreversible, substrate-level phosphorylation





The fate of pyruvate is depends on the type of the cell and on the ability to oxgen.

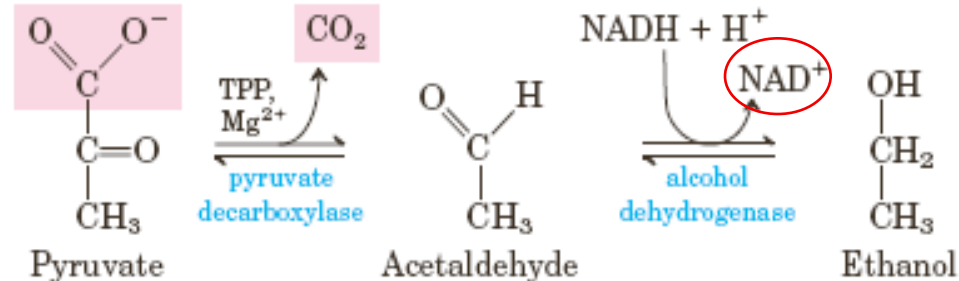
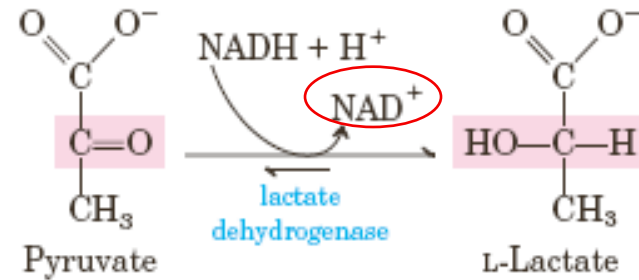
aerobe: pyruvate \longrightarrow acetyl-CoA \longrightarrow TCA cycle

anaerobe: pyruvate \longrightarrow lactate (enzyme: lactate-dehydrogenase)

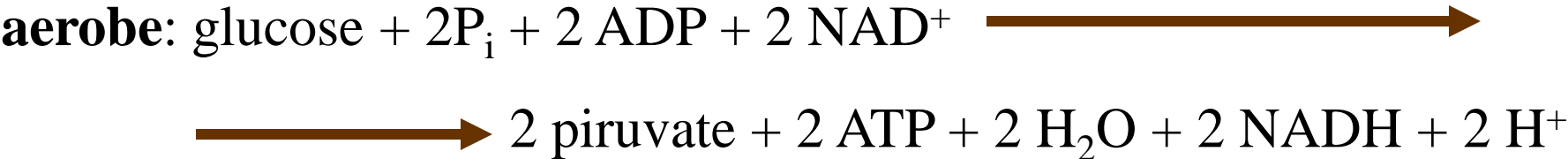
acetaldehyde

ethanol

Alcoholic
fermentation

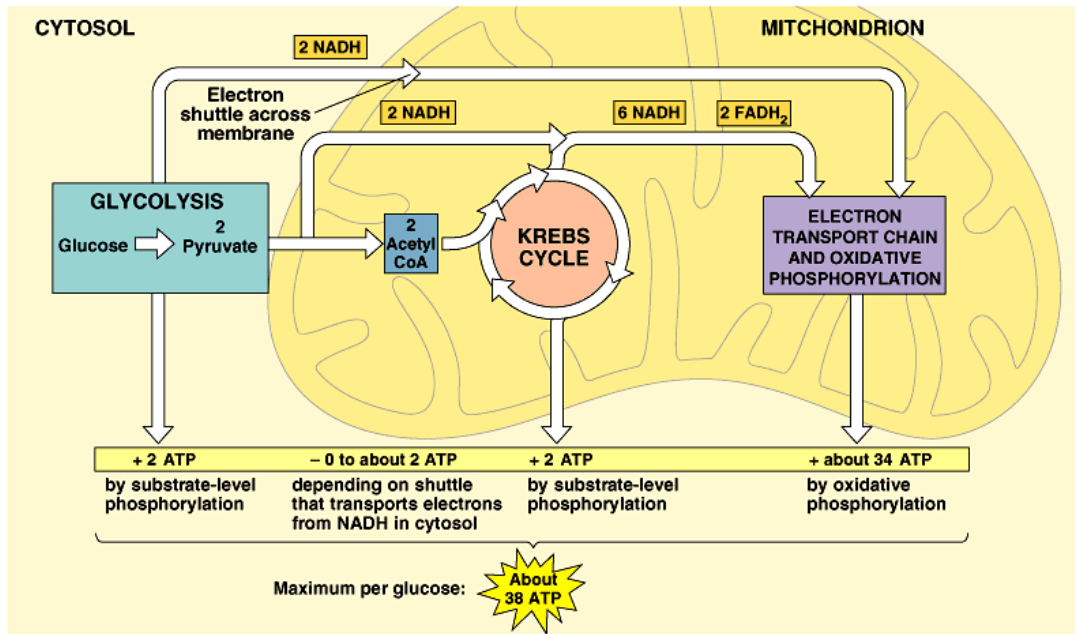
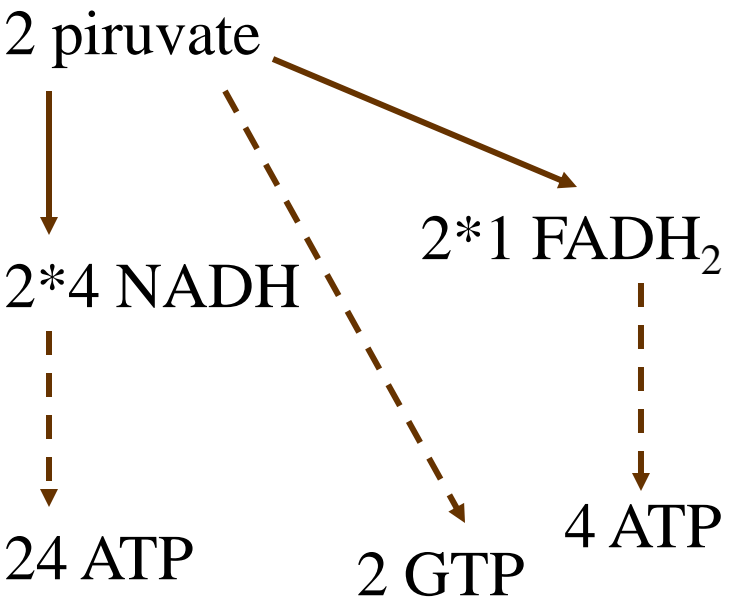


The energy balance of glycolysis



Σ: **36-38 ATP**

2 NADH - - - - -> 4-6 ATP

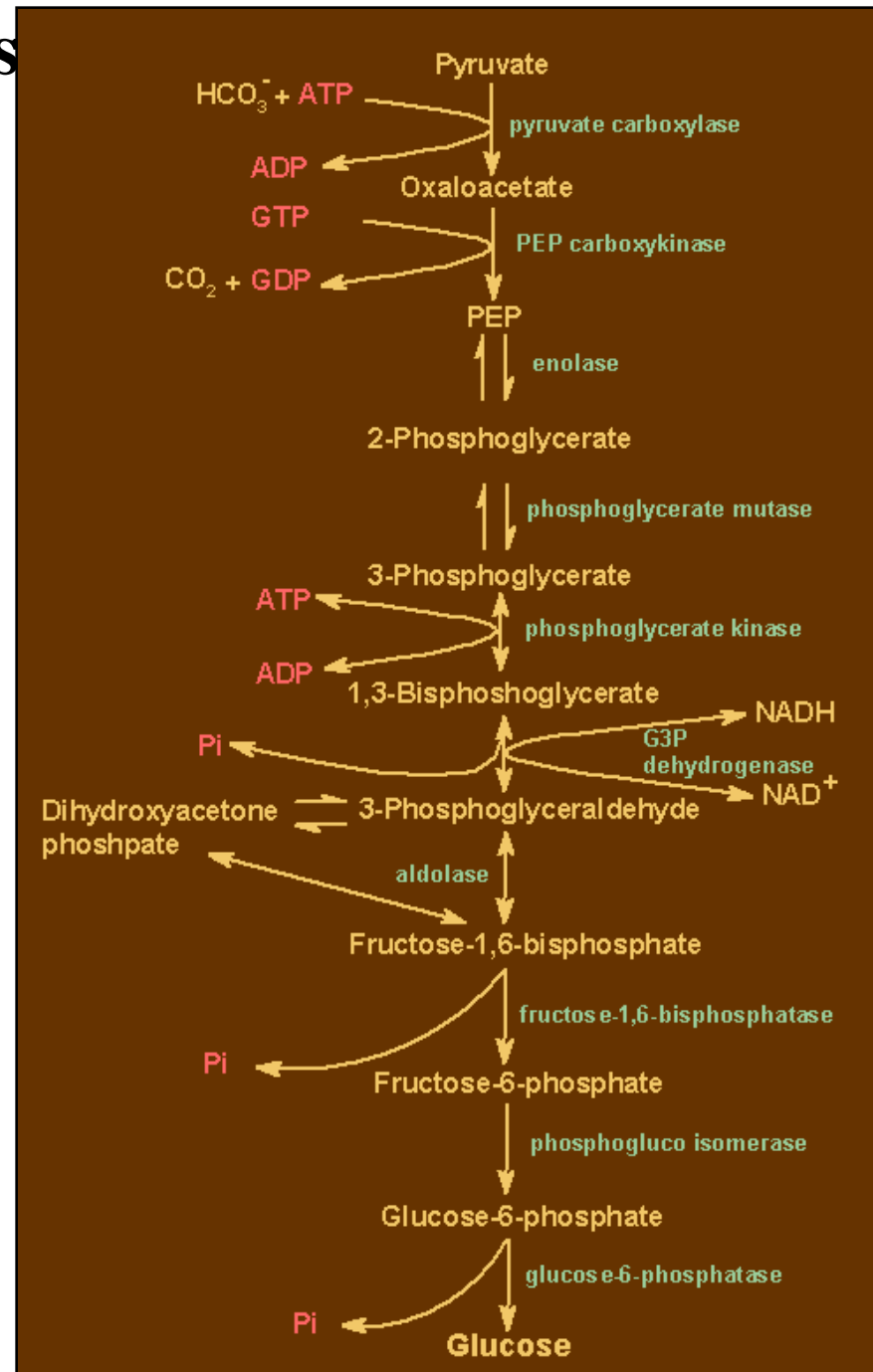


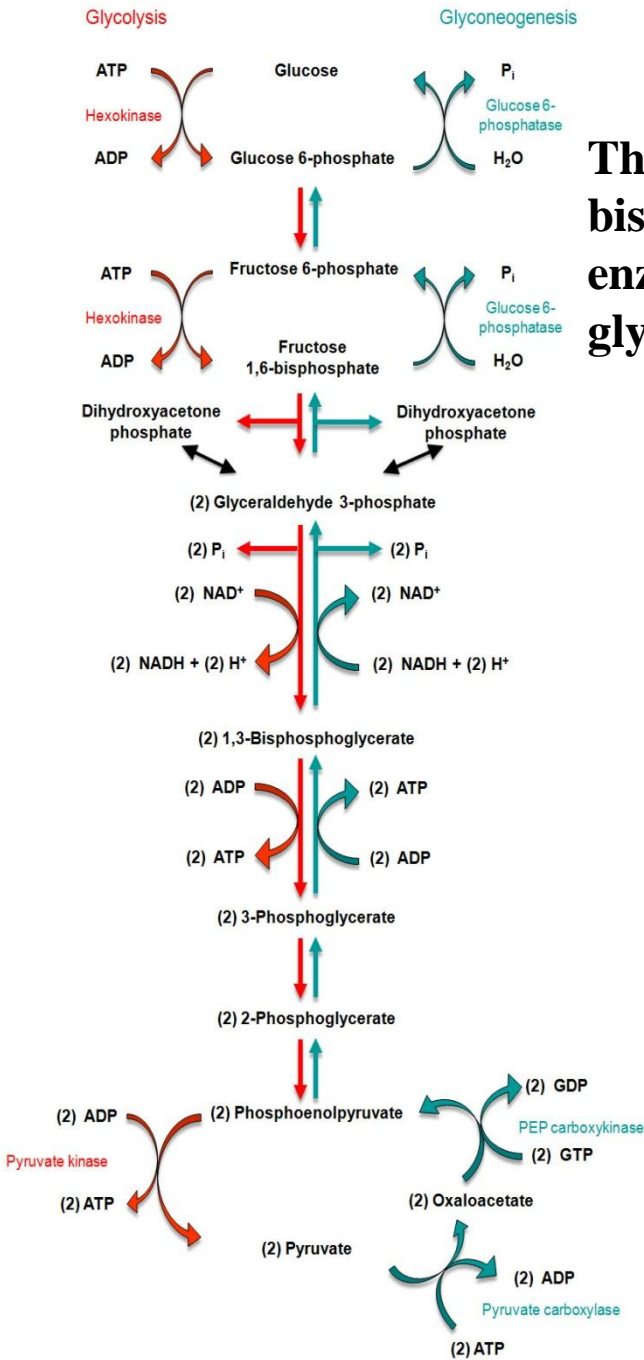
Gluconeogenesis

The process by which glucose is synthesized from noncarbohydrate precursors (eg. lactate), occurs mainly in the liver under fasting conditions.

The reverse of the glycolysis except 3 steps. The exceptions are the irreversible steps (and enzymes catalyze them):

1. Hexokinase
2. Phosphofructokinase-1
3. Pyruvate kinase





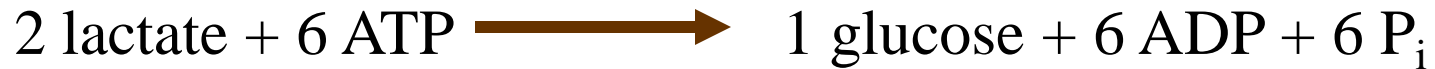
The reactions that remove phosphate from fructose 1,6-bisphosphate and from glucose 6-phosphate each use single enzymes that differ from the corresponding enzymes of glycolysis.

Pyruvate is carboxylated by pyruvate carboxylase to form oxaloacetate.

Oxaloacetate is transported across the mitochondrial membrane as malate or aspartate

Oxaloacetate, produced from malate or aspartate in the cytosol, is converted to PEP by the cytosolic PEP carboxykinase

Energy requirement of gluconeogenesis:



Gluconeogenesis is occurred in the liver and kidney, main organ: liver.

Cori-cycle

